## Structure Reports

Online
ISSN 1600-5368

Run-Hong Jia, ${ }^{\text {a,b }}$ Qian-Shang Zhang, ${ }^{\text {c }}$ Shu-Jiang Tu ${ }^{\text {a }}$ and Yan Zhang ${ }^{\text {a }}$

${ }^{\text {a }}$ Department of Chemistry, Xuzhou Normal University, Xuzhou 221116, People's Republic of China, ${ }^{\mathbf{b}}$ Lianyungang Teachers' College, Lianyungang 222006, People's Republic of China, and ${ }^{\mathbf{c}}$ SuQian Municipal Public Security Bureau, SuQian 223800, People's Republic of China

Correspondence e-mail: laotu2001@263.net

## Key indicators

Single-crystal X-ray study
$T=298 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.006 \AA$
$R$ factor $=0.044$
$w R$ factor $=0.119$
Data-to-parameter ratio $=13.2$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
## 12-(4-Bromophenyl)-9,9-dimethyl-1,2,3,4,9,10-hexahydrobenz[a]acridin-11-one

The title compound, $\mathrm{C}_{25} \mathrm{H}_{22} \mathrm{BrNO}$, has been synthesized by the reaction of 4-bromobenzaldehyde, 3,3-dimethylcyclopentane-1,3-dione with 2-naphthylamine in ethanol. In the crystal structure, the molecules are connected by $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds, forming chains along the $a$ axis.

## Comment

Many natural and synthetic compounds containing the acridine skeleton display interesting biological and physical activities, such as antimalaria (Nasim \& Brychey, 1979; Thull \& Testa, 1994; Reil et al., 1994; Mandi et al., 1994) and antitumour properties (Khurana et al., 1990). Multihydroacridinone derivatives have been reported to have high fluorescence efficiencies and can be used as fluorescent molecular probes for the monitoring of polymerization processes (Popielarz et al., 1997). Increasingly, they also receive attention due to the similarity of their properties with those of 1,4dihydropyridines, which have similarities in structure with biologically important compounds, such as nicotinamide adenine dinucleotide (Srividya et al., 1996). In this paper, we report the crystal structure of the title compound, (I).

(I)

In compound (I), atoms C 7 and N 1 deviate from the $\mathrm{C} 1 / \mathrm{C} 6 /$ C8/C17 plane by 0.321 (5) and 0.151 (5) $\AA$, respectively (Fig. 1), indicating a boat conformation. Atom C3 deviates from the C1/C2/C4/C5/C6 plane by 0.658 (5) A. indicating an envelope conformation. The dihedral angle between the C1/ C6/C8/C17 plane and the C20-C25 benzene ring is $89.98(11)^{\circ}$.

In the crystal structure, the molecules are connected via $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (Table 1), forming chains along the $a$ axis (Fig. 2).

## Experimental

Compound (I) was prepared by the reaction of 4-bromobenzaldehyde ( 1 mmol ) with 3,3-dimethylcyclopentane-1,3-dione

Received 13 April 2006
Accepted 21 April 2006


Figure 1
The molecular structure of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the $30 \%$ probability level.
( 1 mmol ) and 2-naphthylamine $(1 \mathrm{mmol})$ in ethanol $(3 \mathrm{ml})$ at 351 K . Single crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of a $95 \%$ aqueous ethanol solution (yield $95 \%$; m.p. $>573 \mathrm{~K}$ ). Spectroscopic analysis: ${ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}$, $\delta$, p.p.m.): 0.84 $\left(3 \mathrm{H}, s, \mathrm{CH}_{3}\right), 1.04\left(3 \mathrm{H}, s, \mathrm{CH}_{3}\right), 2.03\left(1 \mathrm{H}, d, J=16.4 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 2.23$ $\left(1 \mathrm{H}, d, J=16.4 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 2.39\left(1 \mathrm{H}, d, J=16.8 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 2.56(1 \mathrm{H}, d, J$ $\left.=16.8 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 5.78(1 \mathrm{H}, s, \mathrm{CH}), 7.18(2 \mathrm{H}, d, J=8.4 \mathrm{~Hz}, \mathrm{ArH}), 7.34-$ $7.31(4 \mathrm{H}, m, \mathrm{ArH}), 7.42(1 \mathrm{H}, t, J=7.6 \mathrm{~Hz}, \mathrm{ArH}), 7.82-7.79(2 \mathrm{H}, m$, ArH), $7.90(1 \mathrm{H}, d, J=8.4 \mathrm{~Hz}, \mathrm{ArH}), 9.76(1 \mathrm{H}, s, \mathrm{NH})$.

## Crystal data

$\mathrm{C}_{25} \mathrm{H}_{22} \mathrm{BrNO}$
$M_{r}=432.35$
Triclinic, $P \overline{1}$
$a=7.296(4) \AA \AA$
$b=9.597(5) \AA$
$c=15.304(8) \AA$
$\alpha=94.038(7)^{\circ}$
$\beta=93.465(8)^{\circ}$
$\gamma=103.331(7)^{\circ}$

## Data collection

Bruker SMART CCD area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\text {min }}=0.567, T_{\text {max }}=0.841$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.044$
$w R\left(F^{2}\right)=0.119$
$S=1.01$
3346 reflections
253 parameters
H -atom parameters constrained
$V=1036.9$ (9) $\AA^{3}$
$Z=2$
$D_{x}=1.385 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$\mu=2.00 \mathrm{~mm}^{-1}$
$T=298$ (2) K
Block, colourless $0.32 \times 0.27 \times 0.09 \mathrm{~mm}$

5522 measured reflections 3346 independent reflections 2053 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.023$
$\theta_{\text {max }}=25.0^{\circ}$

$$
\begin{aligned}
& w=1 /[ \sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0454 P)^{2} \\
&+0.7039 P] \\
& \text { where } P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.60 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.48 \mathrm{e} \AA^{-3}
\end{aligned}
$$



Figure 2
A packing diagram for (I), projected along the $a$ axis.

Table 1
Hydrogen-bond geometry $\left(\AA,{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 1-\mathrm{H} 1 \cdots \mathrm{O}^{\mathrm{i}}$ | 0.86 | 1.99 | $2.840(4)$ | 172 |

Symmetry code: (i) $x-1, y, z$.
All H atoms were positioned geometrically and treated as riding, with $\mathrm{C}-\mathrm{H}$ distances in the range $0.93-0.98 \AA$, and with $U_{\text {iso }}(\mathrm{H})=$ $1.5 U_{\text {eq }}(\mathrm{C})$ for methyl H atoms and $1.2 U_{\text {eq }}(\mathrm{C})$ for others.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

The authors thank the National Natural Science Foundation of China (grant No. 20372057), the Natural Science Foundation of Jiangsu Province (grant No. BK2001142) and the Key Laboratory of Biotechnology for Medicinal Plants of Jiangsu Province (grant No. 01AXL 14) for financial support.

## References

Bruker (1998). SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
Bruker (1999). SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
Khurana, J. M., Maikap, G. C. \& Mehta, S. (1990). Synthesis, 8, 731-732.
Mandi, Y., Regely, K., Ocsovszky, I., Barbe, J., Galy, J. P. \& Molnar, J. (1994). Anticancer Res. 14, 2633-2636.
Nasim, A. \& Brychey, T. (1979). Mutat. Res. 65, 261-288.
Popielarz, R., Hu, S. K. \& Neckers, D. C. (1997). J. Photochem. Photobiol. A, 110, 79-83.
Reil, E., Scoll, M., Masson, K. \& Oettmeier, W. (1994). Biochem. Soc. Trans. 22, 62.
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Srividya, N., Ramamurthy, P., Shanmuguasundaramn, P. \& Ramakrishnan, V. T. (1996). J. Org. Chem. 61, 5083-5089.

Thull, U. \& Testa, B. (1994). Biochem. Pharmacol. 447, 2307-2310.


[^0]:    (C) 2006 International Union of Crystallography All rights reserved

